### MEIS studies of oxygen plasma cleaning of copper for fast response time photocathodes used in accelerator applications

<u>Tim Noakes</u><sup>1</sup>, Sonal Mistry<sup>2</sup>, Michael Cropper<sup>2</sup>, **Andrew Rossall**<sup>3</sup>, Jaap van den Berg<sup>3</sup>

 <sup>1</sup> STFC Daresbury Laboratory, SciTech Daresbury, Keckwick Lane, Daresbury, Warrington, WA4 4AD, UK
<sup>2</sup> Department of Physics, Loughborough University, Loughborough, LE11 3TU, UK
<sup>3</sup> International Institute for Accelerator Applications, University of Huddersfield, Queensgate, Huddersfield, HD1 3DH, UK



University o

HRDP8 – 9<sup>th</sup> August, 2016 Western University London, Ontario

### Overview

- Motivation
- Experimental Details
  - MEIS Facility
  - Data Analysis
- Effect of plasma treatment parameters on oxide composition
  - Plasma power
  - Treatment time
  - Annealing temperature
- XPS Data comparison with MEIS
- Summary
- Further work



### Photocathodes

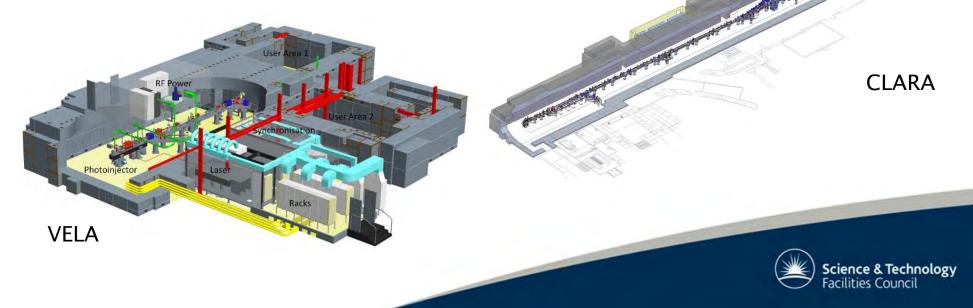
#### Industrial applications

- Photomultiplier tubes
- Image Intensifiers

#### **Accelerator Applications**

- Light sources
- Electron accelerators for high-energy physics

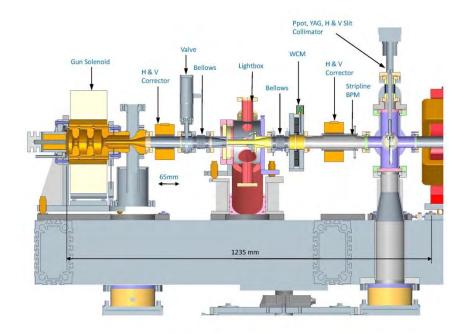
#### Test facilities at Daresbury Laboratory







### 2.5 cell S-band RF Gun



#### 2.5 cell S-band RF gun •

- Cu photocathode:  $QE = 10^{-5}$ ٠
- Sub-100fs mode-locked laser •
- Field gradient of 100 MV/m •
- Max beam energy = 6.5 MeV۲

#### Photocathode preparation

- carried out ex-situ •
- cathode transferred in-air •

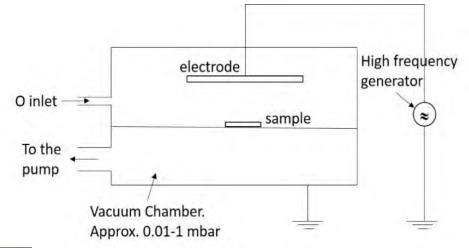




### Cathode Preparation

#### VELA Cu photocathode prepared by:

- O<sub>2</sub> plasma cleaning
  - Removes hydrocarbons
  - Likely to leave a thin protective oxide layer
- Heating to 250 °C (system bake)



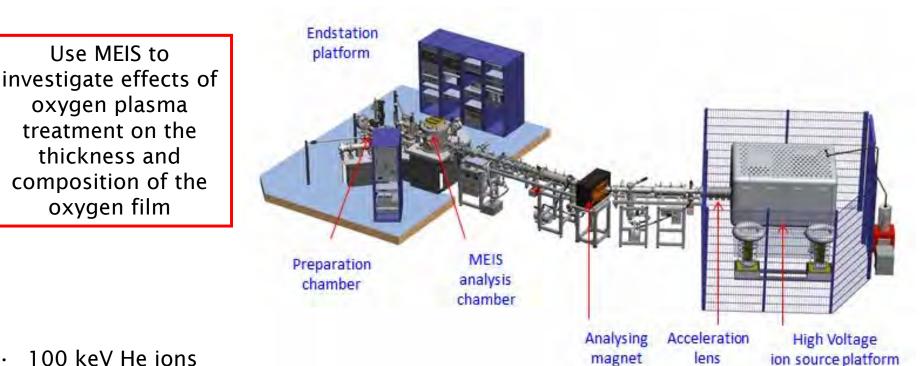
Schematic of plasma cleaner

Poor detailed understanding of the changes in composition and thickness of the oxide film in this preparation process





### MEIS facility at University of Huddersfield



University of HUDDERSFIELD

Science & Technology Facilities Council

- $\cdot$  35.3° incidence angle, 90° scattering angle
- 0.5 x 1.0 mm spot size with a dose of 1.25 µC per tile (1.27 x 10<sup>15</sup> atoms/cm<sup>2</sup>)

### Experimental Detail

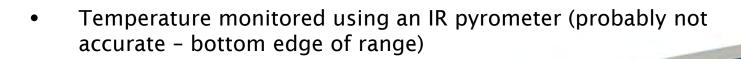
All samples cut from 'as-rolled' oxygen free copper cleaned with acetone and propanol

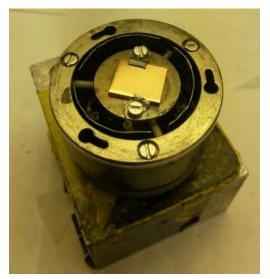
Investigate the Cu surface of samples exposed to oxygen plasma with

- Power levels between 10-100% (20-200W)
- Treatment times between 10-40 minutes

Samples mounted and transferred in-air to MEIS system

- Post heat treatment studies (in-situ MEIS)
  - ~300°C by radiative heating
  - ~600°C by e-beam heating





UDDERSFIE



### MEIS Energy Spectra

Ar sputtered samples are essentially pure Cu at the surface and throughout

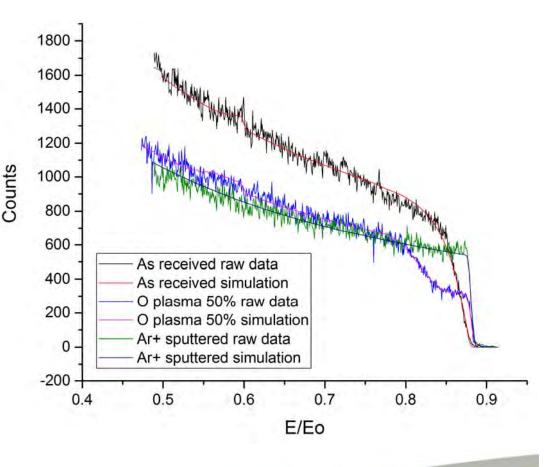
 Allows independent calibration of energy and resolution

As-loaded samples have lower energy and more inclined leading edge

 Thin (~4nm) layer of hydrocarbon contamination

Oxygen treated samples have no hydrocarbon but thick oxide layer has been grown

- Step width gives the thickness
- Height gives composition (O signal not large enough to influence fit significantly)



University of UDDERSFIEL

Science & Technology Facilities Council

# **Depth Profiling**

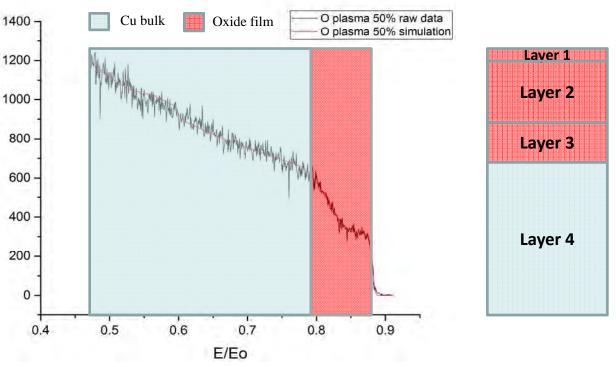
Data fitted using SIMNRA 5.02

#### Layer 1: Surface Layer

- Set to a couple of monolayers thickness (10x10<sup>15</sup> Atoms/cm<sup>2</sup>).
- Composition was treated as variable

#### Layer 2: Oxide Layer

• Composition and the thickness of this layer were treated as variable to best produce a good fit.



University of HUDDERSFIELD

#### Layer 3: Interfacial layer

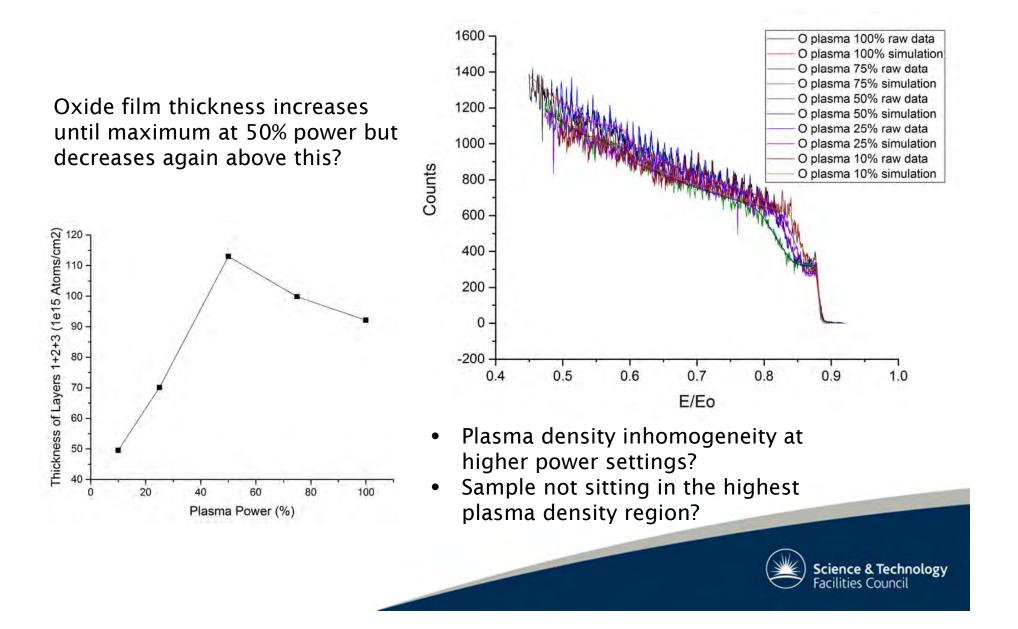
- Composition and the thickness treated as variable
- This layer probably has gradually varying composition, but is modelled by a single composition with roughness added (in layer above) to smear out the signal and improve the fit.

#### Layer 4: Cu Bulk

• Pure Copper bulk

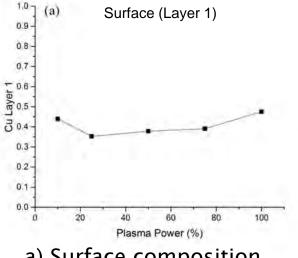


### **Plasma Power Variation**



### Plasma Power Variation

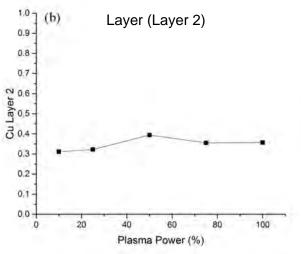
#### Copper fraction as a function of depth and plasma power



a) Surface composition slightly higher than layer beneath (40.7%)

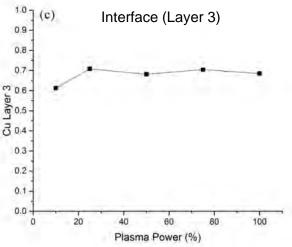
- Anomalous surface peaks seen previously for oxides (charge fraction)\*
- Small reduction in oxygen content?

\* See for example Kido, Nishimura and Fukumura, PRL 82 (1999) 3352



b) Layer composition consistent with CuO<sub>2</sub> (34.8%)

 Unusual form of copper peroxide formed in highly oxidising environments?



University of HUDDERSFIEL

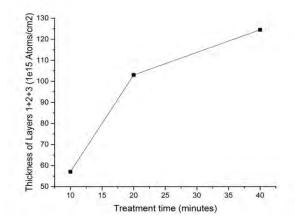
c) Interface composition  $Cu_2O$  (67.8%) on average, but more likely to be gradually changing composition from higher oxygen content to bulk metal



### Treatment time Variation

Film thickness increases with treatment time broadly as expected

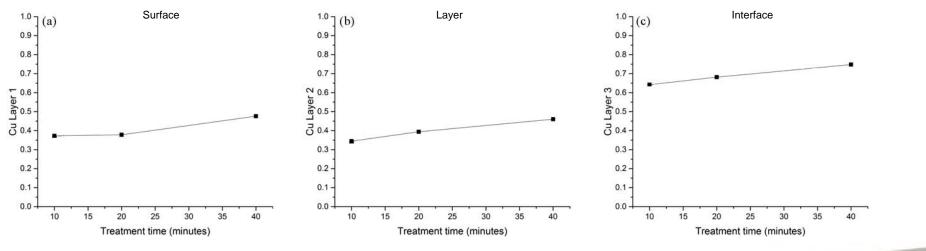
- Initial rapid increase with time
- As thickness increases diffusion and/or penetration of energetic species limits further increase



University of

HUDDERSFIELD

#### Copper fraction as a function of depth and treatment time



Composition data essentially shows the same behaviour as power variation data



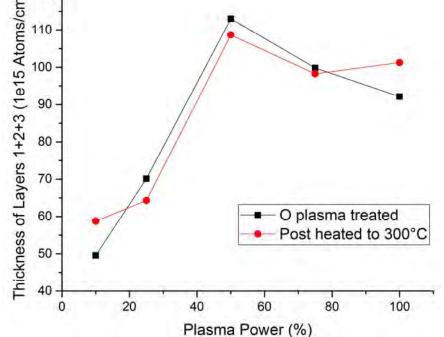
### Annealing

Film thickness unaffected by 300°C 
 Thickness of Layers 1+2+3 (1e15 Atoms/cm2)

 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0
 0 anneal (but completely removed by 600°C treatment) 110 >100 Cu-90 content (%) 80 70 Cu₂0→ 60 Cu 50 CuO - Cu content

 $CuO_2 \longrightarrow \begin{pmatrix} 40 \\ 30 \\ 0 \\ 0 \\ 100 \\ 200 \\ 300 \\ 400 \\ 500 \\ 600 \\ 700 \\ 800 \\ Temperature (°C)$ 

Further data required to accurately assess the minimum temperature for complete oxygen removal



University of HUDDERSFIELD

Film composition moves to an average of 45%, closer to CuO (second layer of the model)



### **XPS** Analysis



#### Depth Profiling with Thermo K-alpha instrument

Al K alpha monochromated source (1486.6 eV)

Flood gun for charge compensation 300 micron spot size

Snap scans with 128 channels, 151.2 eV pass energy, 5 scans at 1s (total time is 5s)

Etching: 200 eV Ar<sup>+</sup> ion beam, 1.2 mm raster, 0.03nm/s based on  $Ta_2O_5$  etch rate.

The peak fitting with a Lorentzian(70%)/ Gaussian(30%) mix and Shirley background.

Scofield Relative Sensitivity Factors for quantification.

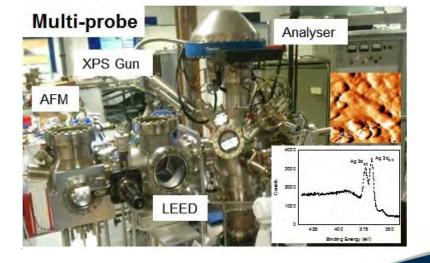
### High Resolution spectra from Multiprobe instrument at Daresbury

Mg K alpha source (1253.6 eV)

Alpha110 analyser with 1.1 eV resolution average of 5 scans

CasaXPS analysis (Shirley background, Scofield RSFs)

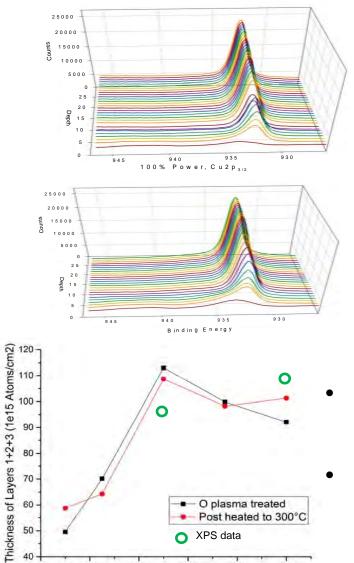




# XPS Depth Profile Cu 2p<sub>3/2</sub>



50% Power, Cu2p<sub>3/2</sub>



0

0

20

40

- O plasma treated Post heated to 300°C

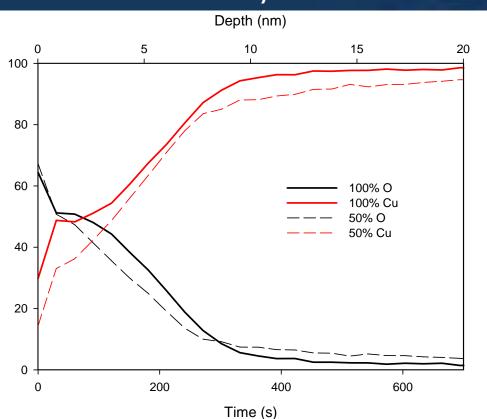
80

100

XPS data

60

Plasma Power (%)



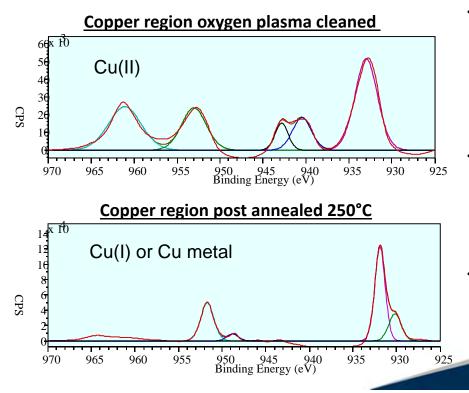
- XPS sputter depth profiling for 50 and 100% power shows similar depths although 100% now marginally thicker.
  - XPS data is consistent with previous MEIS analyses



# High Resolution XPS Data

Compositional analysis shows very high oxygen content

- Hydrocarbon contamination in as received is confirmed
- O<sub>2</sub> plasma treatment removes a significant amount (~95%) of this contamination



Metal	QE	Cu 2p <sub>3/2</sub>	XPS (%) O 1s	C 1s
<b>Cu</b> Received O <sub>2</sub> plasma Anneal 250°C ½ hrs	4.2E-6 0 1.6E-4	0.6 16.8 16.8	40.4 80.1 76.7	59.0 3.1 6.5

- Chemical shift data shows Cu(II) for the plasma treated sample ( $CuO_2$  and CuO both have copper in the 2+ oxidation state)
- However, annealed sample indicates Cu(I) or metallic Cu, not consistent with compositional analysis or MEIS data?
- QE after anneal: 1-2 orders up



### Summary

Oxygen plasma treatment removes hydrocarbon contamination and leaves an oxygen rich film  $(CuO_2!)$ 

- Increasing treatment time increases thickness; composition remains the same
- Increasing power increases thickness until 50% above which it may be position sensitive?

Annealing to 250-300°C changes composition (less oxygen), however oxide thickness remains unchanged

- Doesn't appear to give metallic surface
- Oxide surfaces have higher work function
- Additional 'cleaning' from high power UV photoinjector laser during operation (QE of 2×10<sup>-5</sup> seen)?

XPS results are largely consistent with the MEIS analysis

- Similar depth profiles and precise composition with the exception of an annealed sample
- XPS does agree with the high oxygen content
- Indicates the necessity for further work investigate the effect of the annealing process on the precise composition



University of

UDDERSFIE

### Further Work

More detailed evaluation of annealing behaviour

- More accurate temperature measurement (thermocouple)
- More data points at different temperatures

Single crystal studies to determine the effect of surface orientation

• (111), (110) and (100) surfaces



### Acknowledgments

**Pat Cropper** - Loughborough Materials Characterisation Centre

# **Reza Valizadeh, Adrian Hannah** – ASTeC Vacuum Group

**Boris Militsyn, Elaine Seddon** – ASTeC Accelerator Physics Group



University (